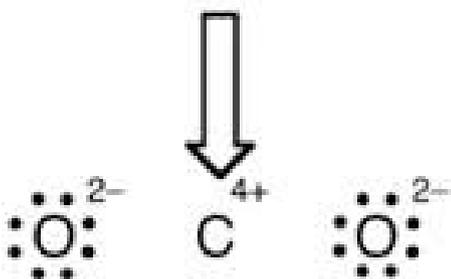
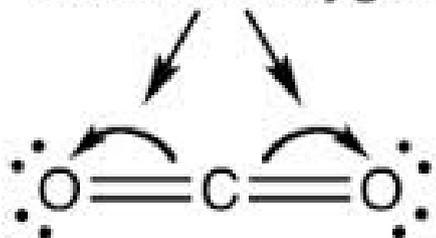


# Carica fomale vs numero ossidazione

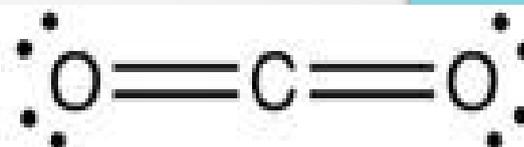
"Award" electrons in bonds to oxygen



NUMERO OSSIDAZIONE

Formal Charge

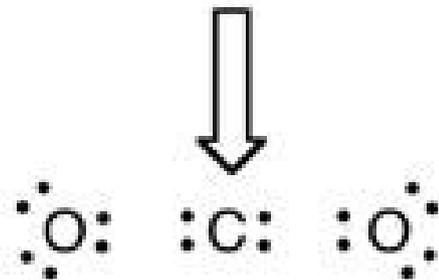
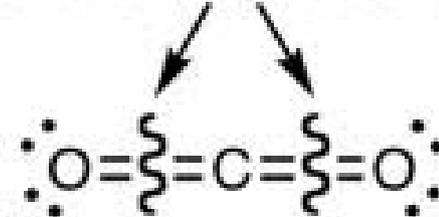
Oxidation State



0      0      0

2-      4+      2-

"Split" electrons in covalent bonds evenly



CARICA FROMALE

# Nomenclatura

$\text{ClO}^-$  ipoclorito

+1

$\text{ClO}_2^-$  clorito

+3

$\text{ClO}_3^-$  clorato

+5

$\text{ClO}_4^-$  perclorato

+7